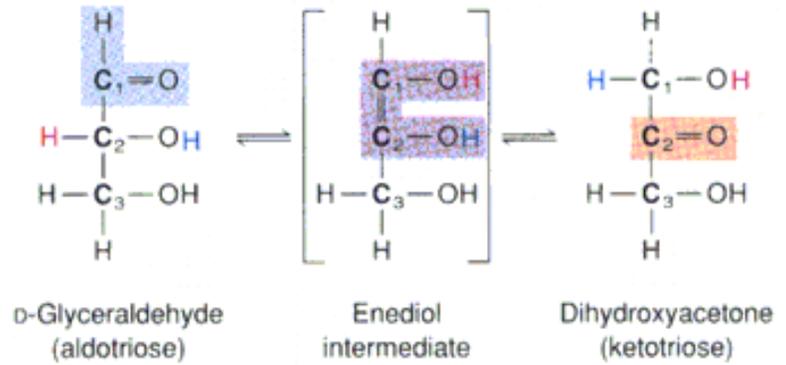
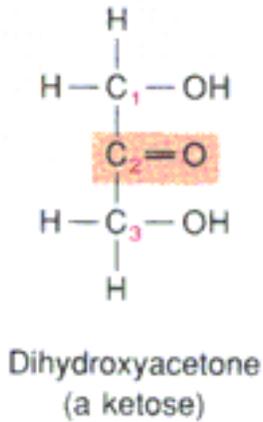
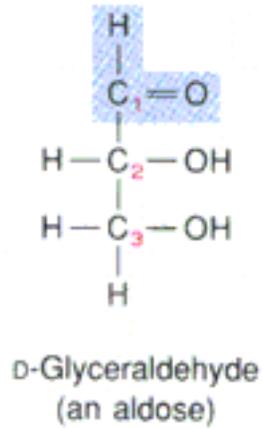


Hidratos de Carbono

Características generales

- Fórmula general (CH_2O)
- Monosacáridos, Oligosacáridos, Polisacáridos
- Glicoproteínas, glicolípidos, glicolipoproteínas y ácidos nucleicos
- Importancia metabólica, estructural, información
- Polihidroxialdehídos y polihidroxicetonas
- “osa”, por ejemplo aldotriosa, cetotriosa

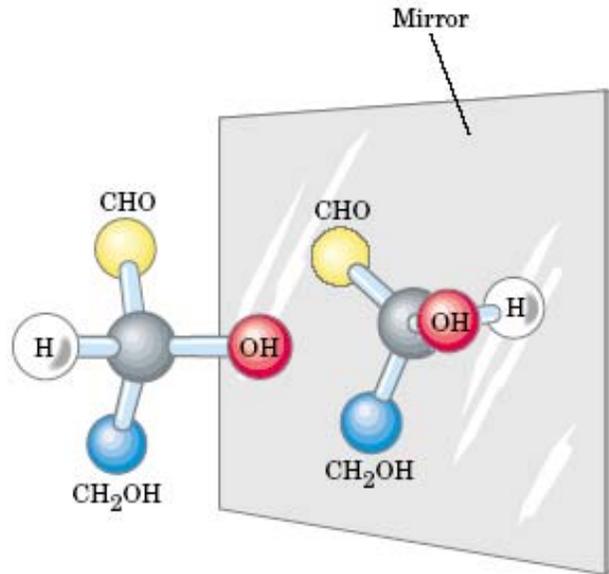


Condiciones débilmente alcalinas

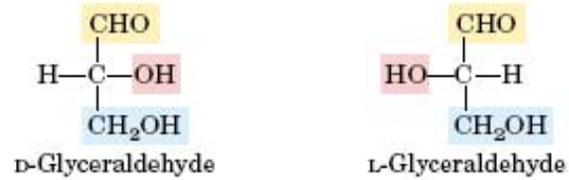
Monosacáridos

- Sólidos cristalinos
- Solubles en agua, incoloros
- Insolubles en solventes orgánicos
- Desvían el plano de la luz polarizada (actividad óptica), moléculas asimétricas
- inestables en medio ácido fuerte (derivados furánicos)

Carbonos asimétricos e isomería



Ball-and-stick models



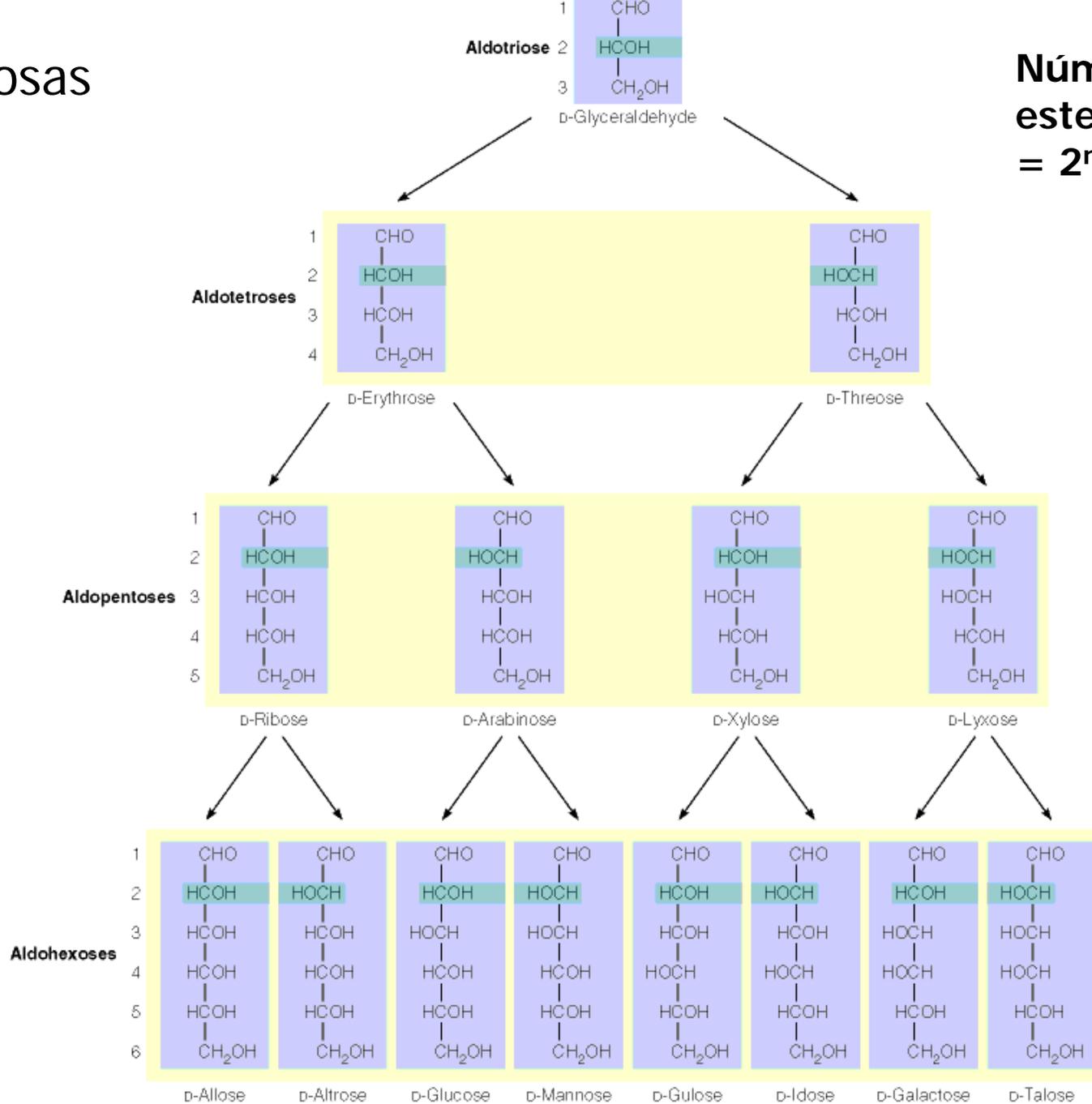
Fischer projection formulas



Perspective formulas

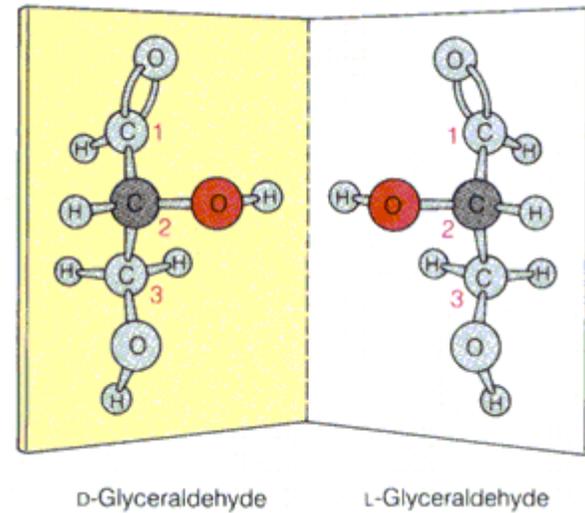
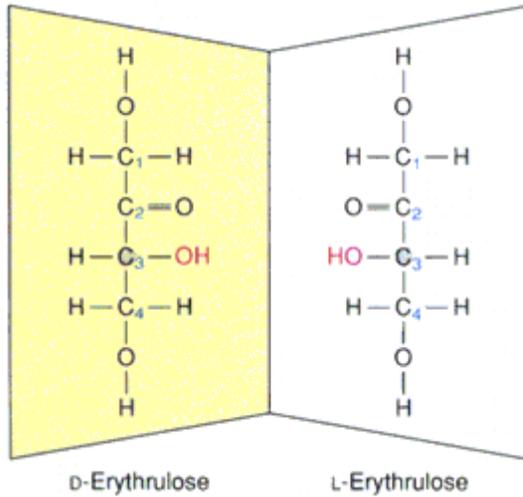
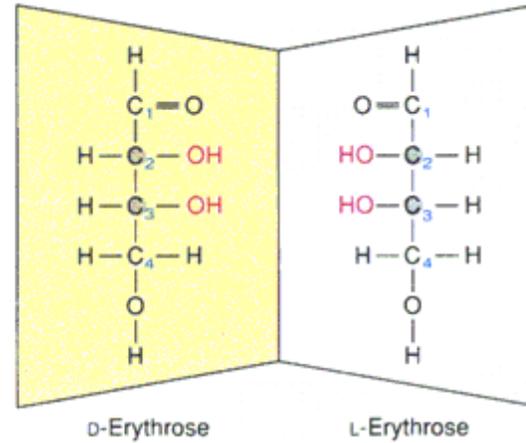
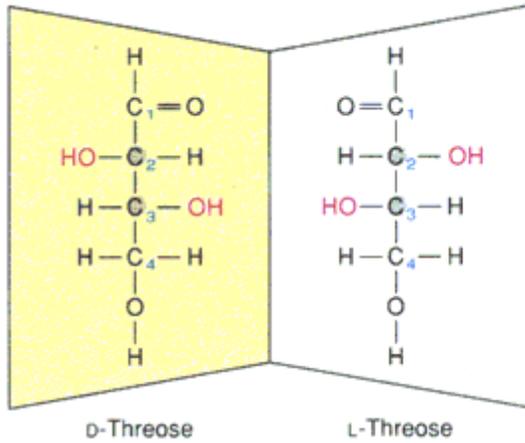
Aldosas

Número de estereoisómeros = 2^n



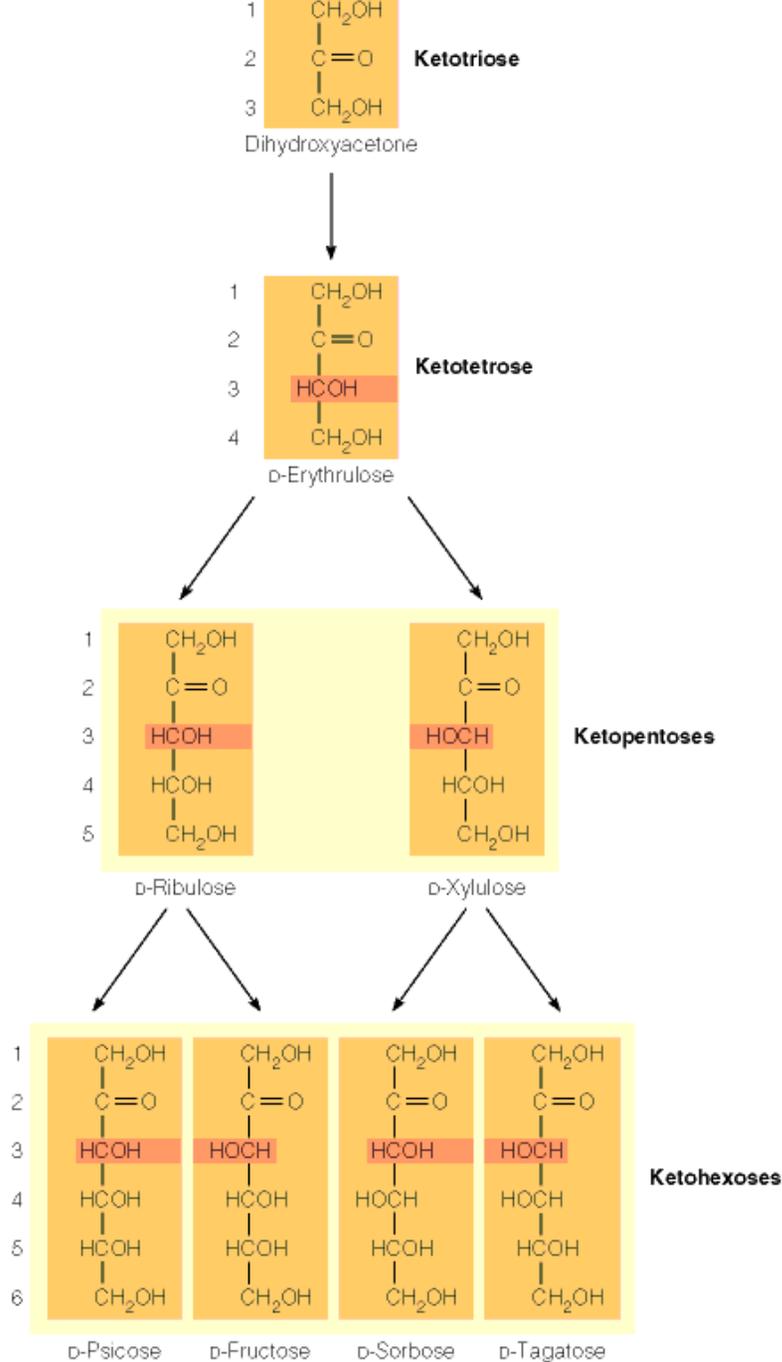
(a) D-Aldoses

Enantiómeros y diastereoisómeros



Cetosas

Número de estereoisómeros = 2^n

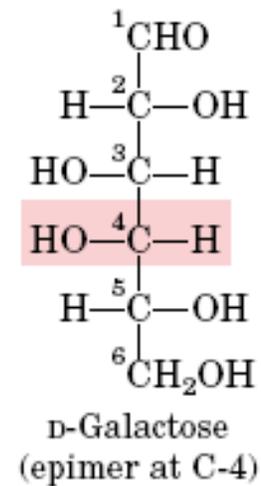
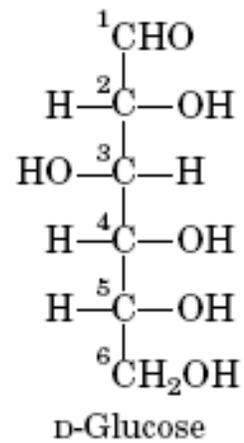
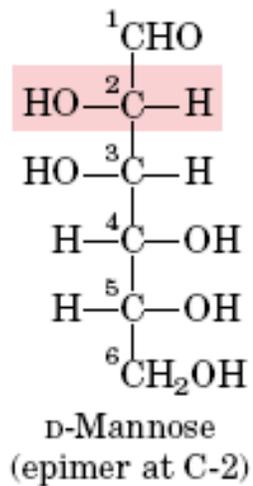


(b) D-Ketoses

Monosaccharides	Natural Occurrence	Physiological Role ^a
Trioses		
Glyceraldehyde	Widespread (as phosphate)	The 3-phosphate is an intermediate in glycolysis
Dihydroxyacetone	Widespread (as phosphate)	The 1-phosphate is an intermediate in glycolysis
Tetroses		
D-Erythrose	Widespread	The 4-phosphate is an intermediate in carbohydrate metabolism
Pentoses		
D-Arabinose	Some plants, tuberculosis bacilli	Plant glycosides, cell walls
L-Arabinose	Widely distributed in plants, bacterial cell walls	Constituent of cell walls, plant glycoproteins
D-Ribose	Widespread, in all organisms	Constituent of ribonucleic acid
2-D-Deoxyribose	Widespread, in all organisms	Constituent of deoxyribonucleic acid
D-Xylose	Woody materials	Constituent of plant polysaccharides
Hexoses		
D-Galactose	Widespread	Milk (as part of lactose); structural polysaccharides
L-Galactose	Agar, other polysaccharides	Polysaccharide structures
D-Glucose	Widespread	A major energy source for animal metabolism; structural role in cellulose
D-Mannose	Plant polysaccharides, animal glycoproteins	Polysaccharide structures
D-Fructose	A major plant sugar; part of sucrose	Intermediate in glycolysis (phosphate esters)
Heptoses		
D-Sedoheptulose	Many plants	Intermediate in Calvin cycle in photosynthesis and pentose phosphate pathway

^aSome of these monosaccharides have additional roles that are not listed.

Epímeros



Comportamiento de la glucosa en solución

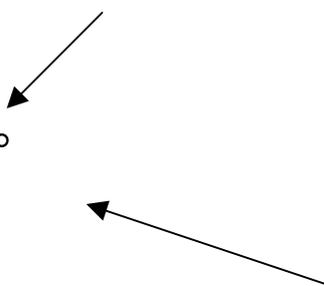
La glucosa en solución no da la reacción de Schiff

La D (+) glucosa metilada presenta dos isómeros

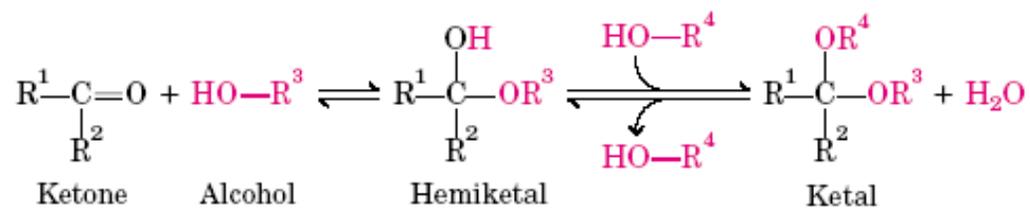
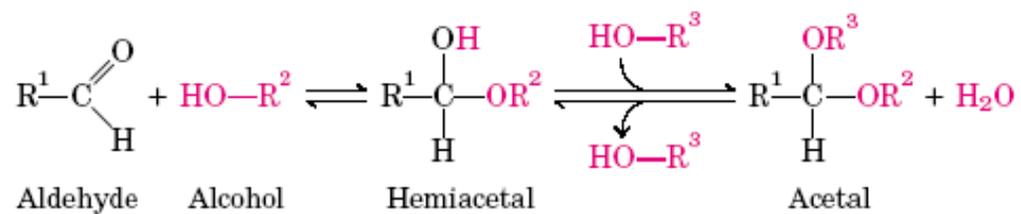
La D glucosa en solución tiene dos formas isoméricas que sufren "mutorotación"

D (+) glucosa cristalina = +112 °

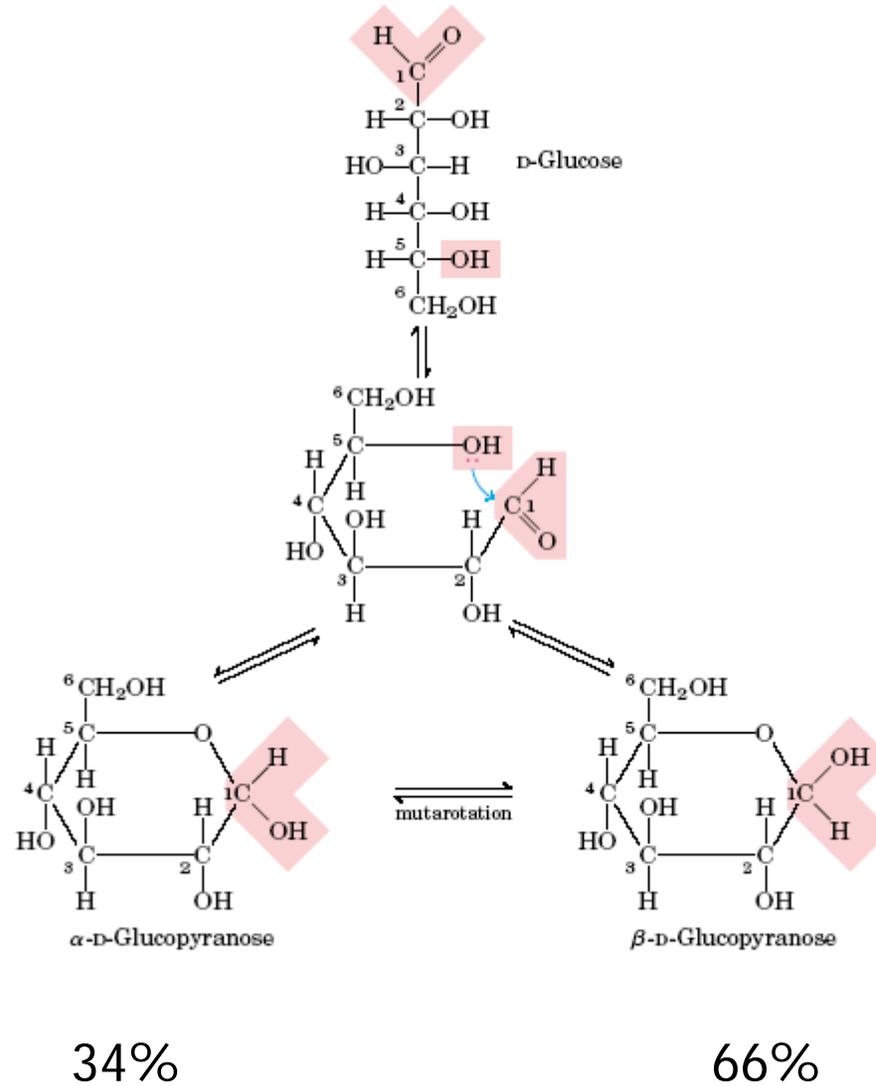
52,7 °



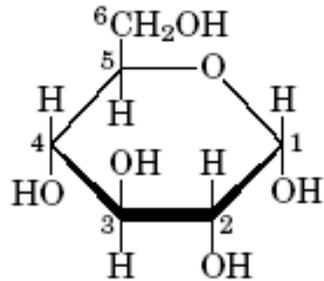
D (+) glucosa cristalis ontenida a 98 °C = +19 °



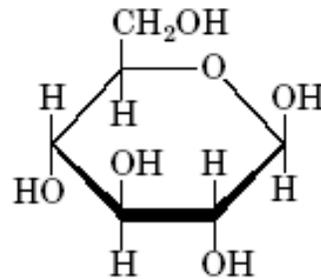
Formas cíclicas



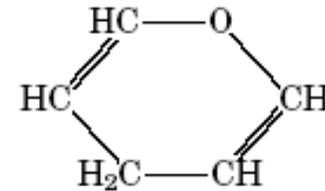
Anómeros



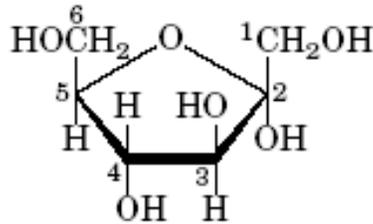
α -D-Glucopyranose



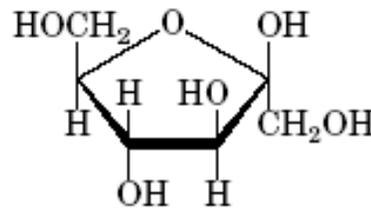
β -D-Glucopyranose



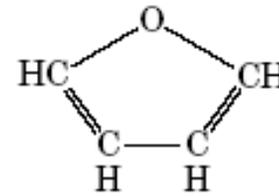
Pyran



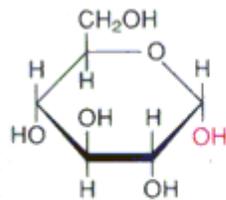
α -D-Fructofuranose



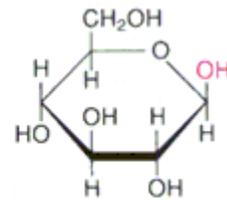
β -D-Fructofuranose



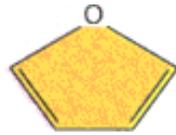
Furan



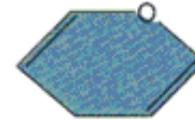
α -D-Glucopyranose



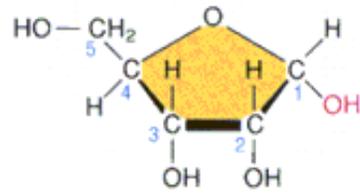
β -D-Glucopyranose



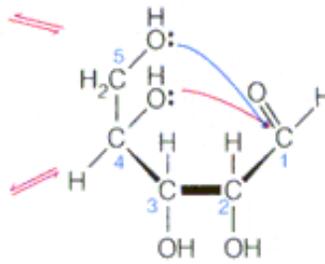
Furan



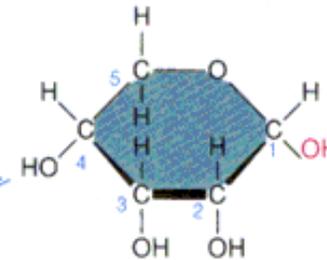
Pyran



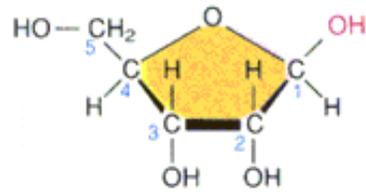
α -D-Ribofuranose



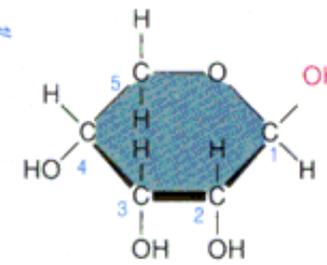
D-Ribose



α -D-Ribopyranose



β -D-Ribofuranose

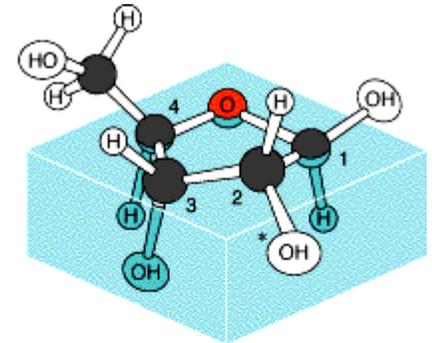
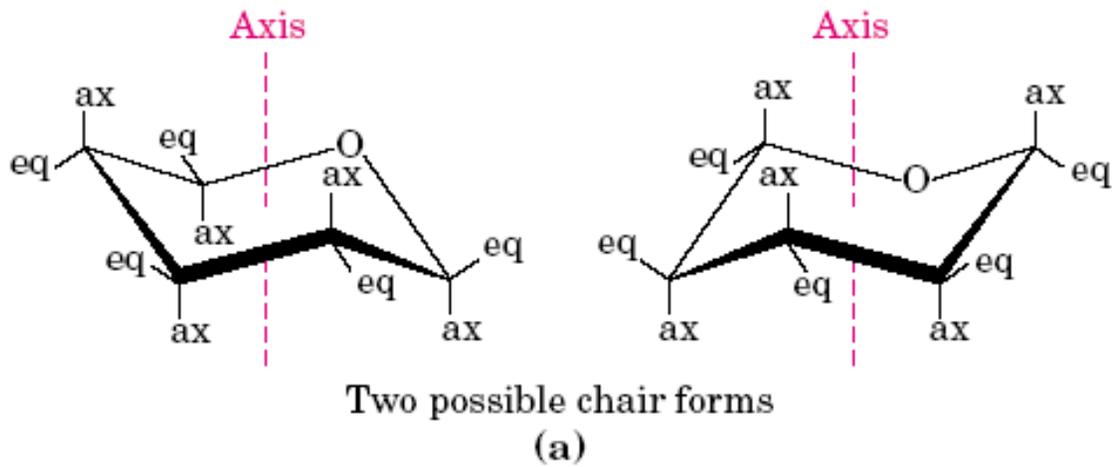


β -D-Ribopyranose

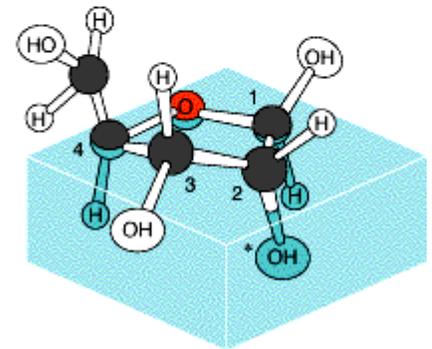
D glucosa en solución

α -piranosa	38,8%
β -piranosa	60,9%
α -furanosa	0.14%
β -furanosa	0.15%
Forma abierta	0.0045%

Isómeros conformacionales



(a) β -D-Ribofuranose, C-2 endo



(b) β -D-Ribofuranose, C-3 endo

Monosaccharide	Relative Amount (%)				Total Furanose
	α -Pyranose	β -Pyranose	α -Furanose	β -Furanose	
Ribose	20	56	6	18	24
Lyxose	71	29	— ^a	— ^a	<1
Altrose	27	40	20	13	33
Glucose	36	64	— ^a	— ^a	<1
Mannose	67	33	— ^a	— ^a	<1
Fructose	3	57	9	31	40

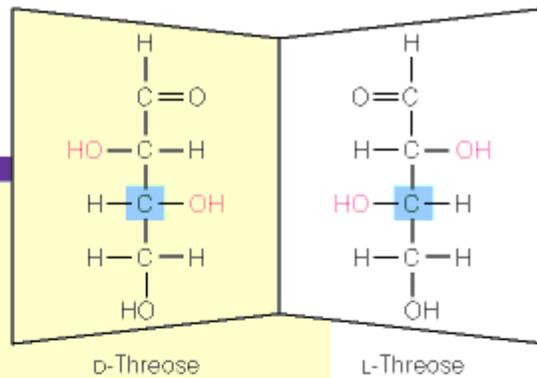
Note: In all cases, the open-chain form is much less than 1%. For data on other sugars, see S. J. Angyal, The composition and conformation of sugars in solution, *Angew. Chem.* (1969) 8:157–226.

^aMuch less than 1%.

Enantiomers

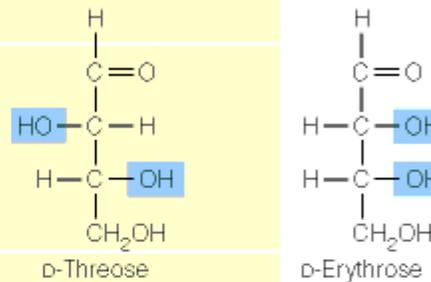
Stereoisomers that are mirror images of one another

The boxed asymmetric carbon (farthest from aldehyde) determines D/L designation



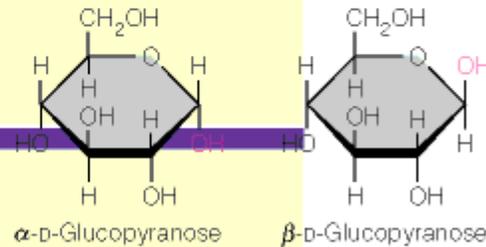
Diastereomers

Stereoisomers that are not mirror images of one another



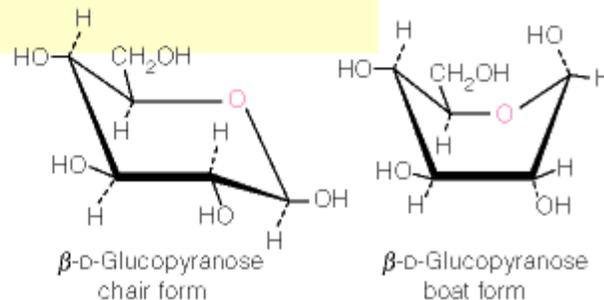
Anomers

Stereoisomers that differ in configuration at the anomeric carbon

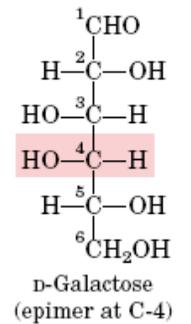
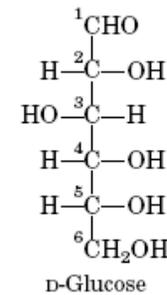
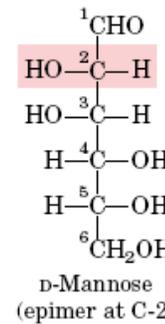


Conformational isomers

Molecules with the same stereochemical configuration, but differing in three-dimensional conformation

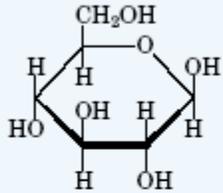


Epimeros

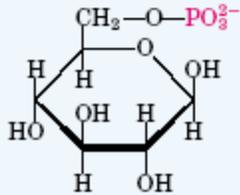


Derivados

Glucose family

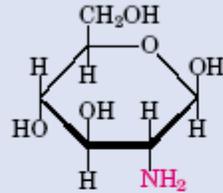


β-D-Glucose

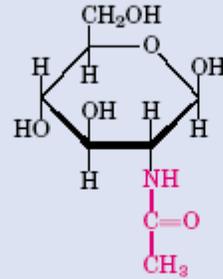


β-D-Glucose 6-phosphate

aminados



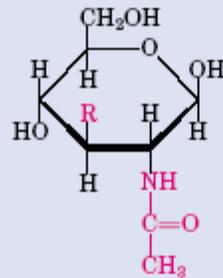
β-D-Glucosamine



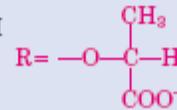
N-Acetyl-*β*-D-glucosamine



Muramic acid



N-Acetylmuramic acid



Amino sugars

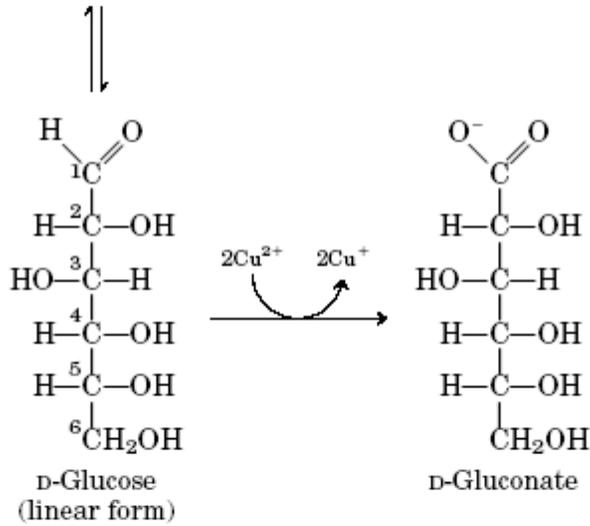
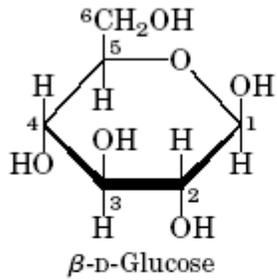


β-D-Galactosamine



β-D-Mannosamine

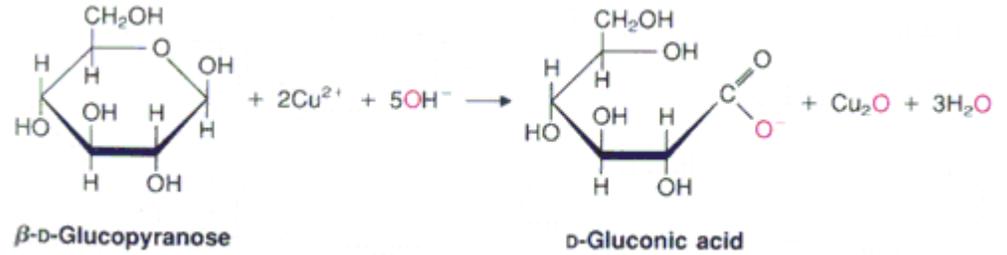
Derivados oxidados



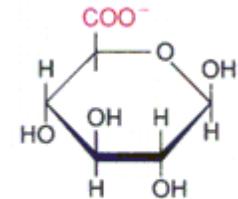
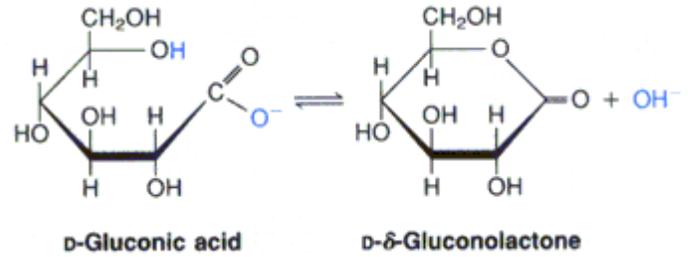
(a)



(b)

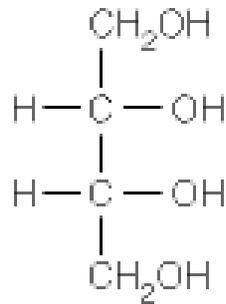


Ac. aldónicos

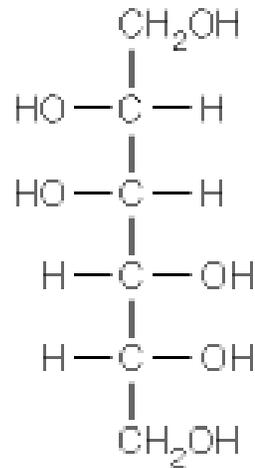


Ac. urónicos

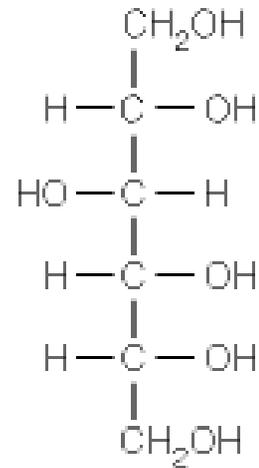
Derivados reducidos (alditoles)



Erythritol*

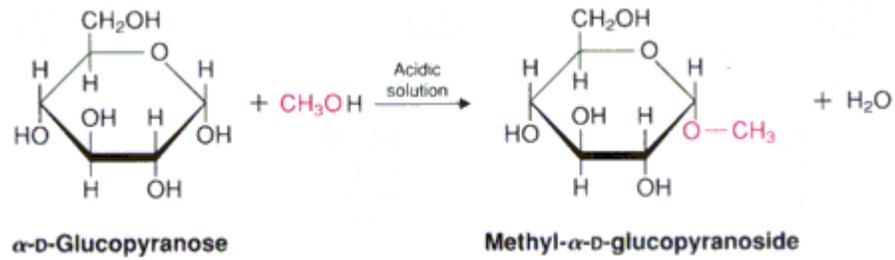


D-Mannitol

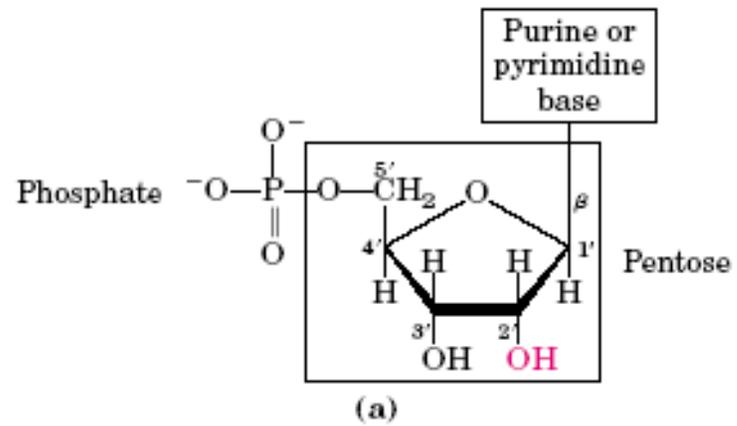
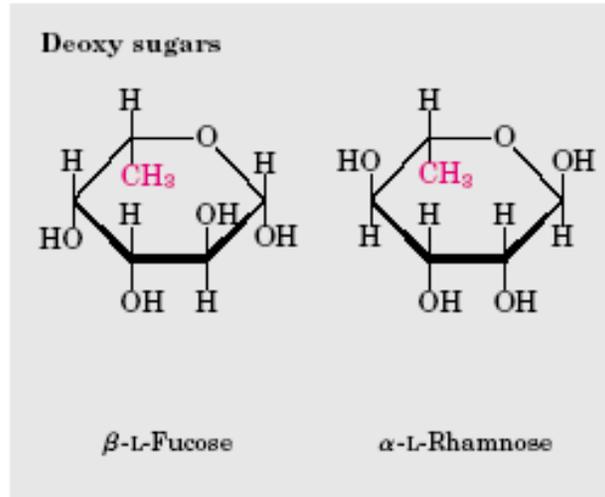


**D-Glucitol
(sorbitol)**

Derivados metilados

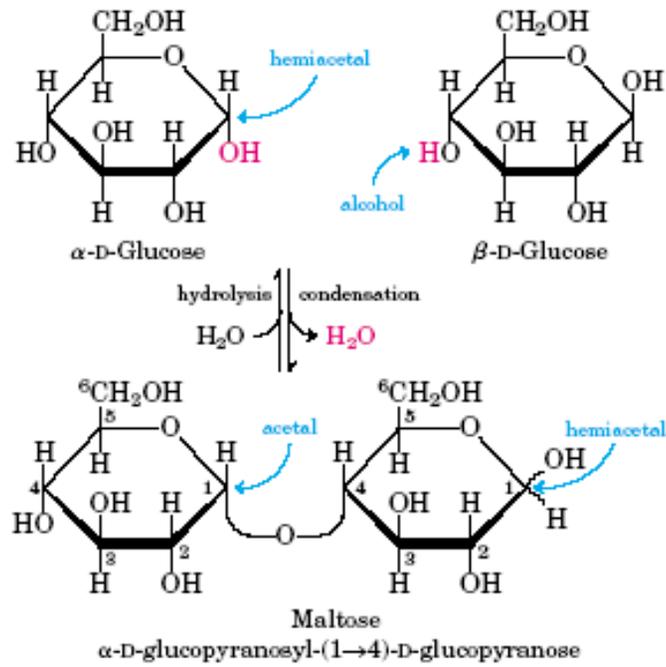


Derivados por sustitución de OH



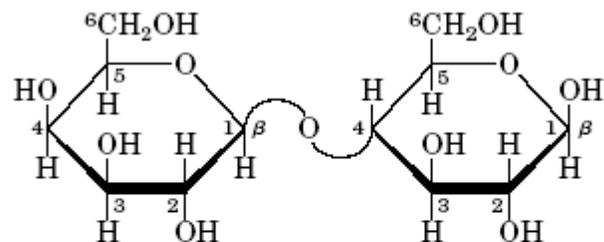
Disacáridos

- Por hidrólisis ácida o enzimática dan monosacáridos
- La unión se denomina glicosídica
- Esta unión es inestable en medio ácido pero relativamente estable en medio alcalino
- Pueden tener o no poder reductor (disacáridos que tienen un grupo hemiacetálico libre (reductores lactosa, maltosa, celobiosa, no reductores sacarosa, trehalosa)
- son sólidos , solubles en agua.
- Tienen actividad óptica

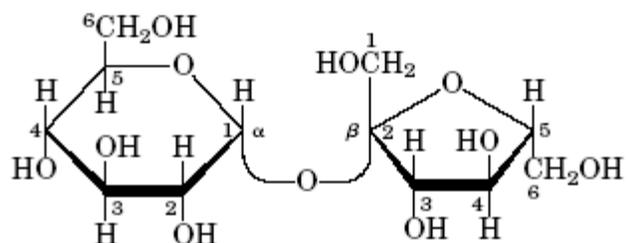


--- --

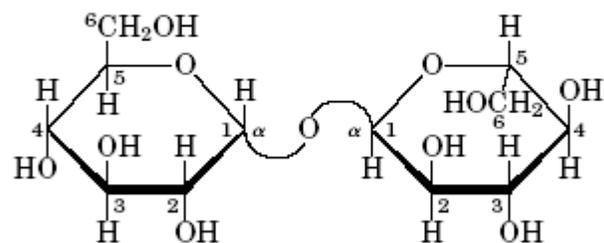
Disaccharide	Structure	Natural Occurrence	Physiological Role
Sucrose	$\text{Glc}\alpha(1\rightarrow2)\text{Fru}\beta$	Many fruits, seeds, roots, honey	A final product of photosynthesis; used as primary energy source in many organisms
Lactose	$\text{Gal}\beta(1\rightarrow4)\text{Glc}$	Milk, some plant sources	A major animal energy source
α,α -Trehalose	$\text{Glc}\alpha(1\rightarrow1)\text{Glc}\alpha$	Yeast, other fungi, insect blood	A major circulatory sugar in insects; used for energy
Maltose	$\text{Glc}\alpha(1\rightarrow4)\text{Glc}$	Plants (starch) and animals (glycogen)	The dimer derived from the starch and glycogen polymers
Cellobiose	$\text{Glc}\beta(1\rightarrow4)\text{Glc}$	Plants (cellulose)	The dimer of the cellulose polymer
Gentiobiose	$\text{Glc}\beta(1\rightarrow6)\text{Glc}$	Some plants (e.g., gentians)	Constituent of plant glycosides and some polysaccharides



Lactose (β form)
 β -D-galactopyranosyl-(1 \rightarrow 4)- β -D-glucopyranose
 Gal(β 1 \rightarrow 4)Glc



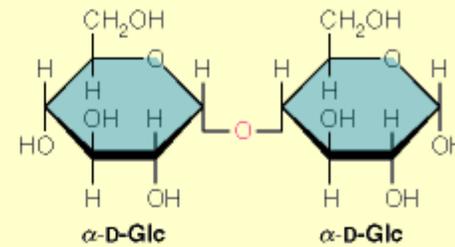
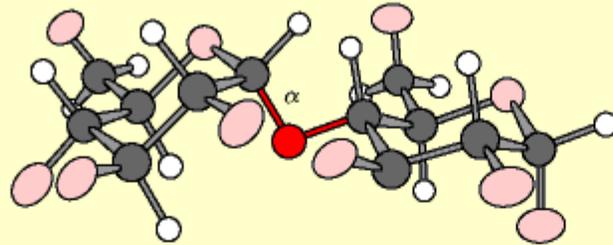
Sucrose
 β -D-fructofuranosyl α -D-glucopyranoside
 Fru(β 2 \leftrightarrow 1 α)Glc



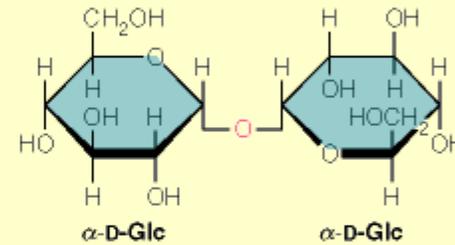
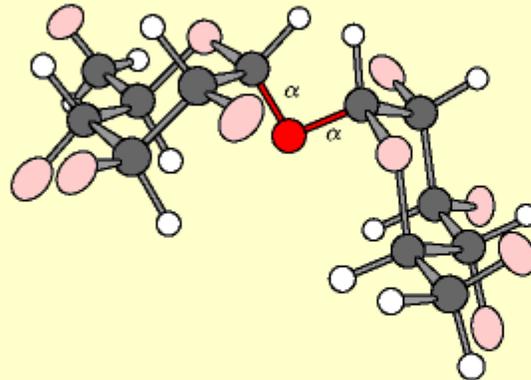
Trehalose
 α -D-glucopyranosyl α -D-glucopyranoside
 Glc(α 1 \leftrightarrow 1 α)Glc

(a) DISACCHARIDES with α connections

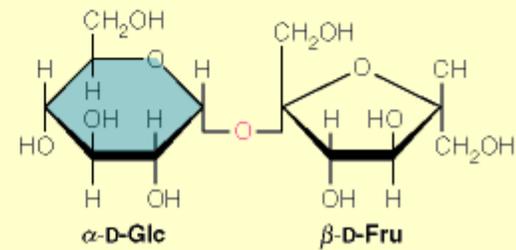
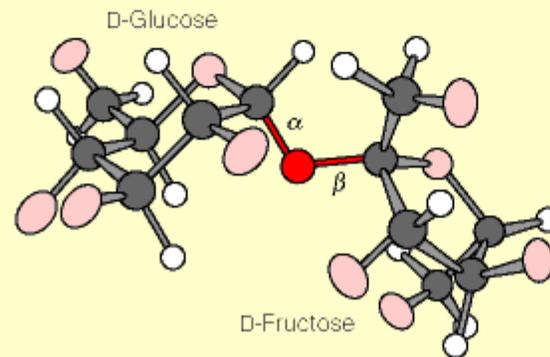
Maltose:
 α -D-glucopyranosyl
(1 \rightarrow 4) α -D-glucopyranose



α,α -Trehalose:
 α -D-glucopyranosyl
(1 \rightarrow 1) α -D-glucopyranose

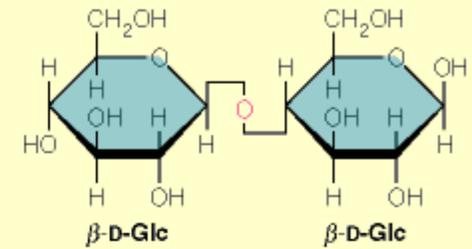
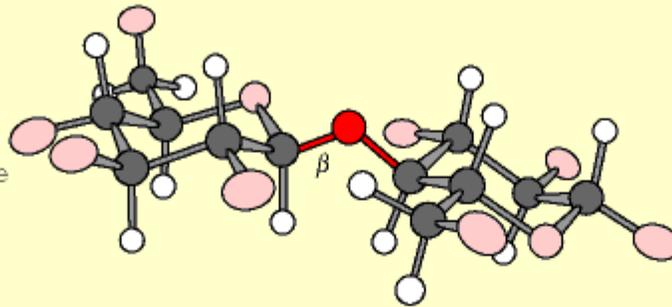


Sucrose:
 α -D-glucopyranosyl
(1 \rightarrow 2) β -D-fructofuranoside

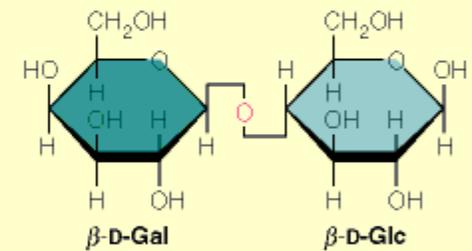
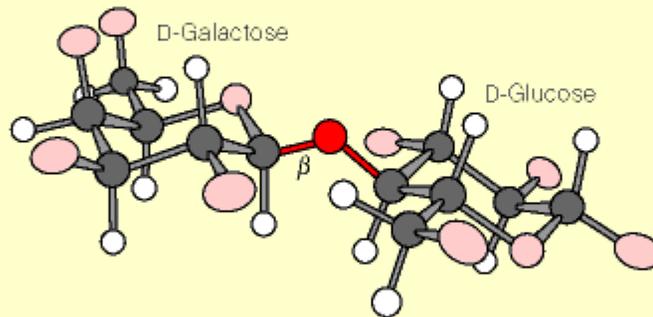


(b) DISACCHARIDES with β connections

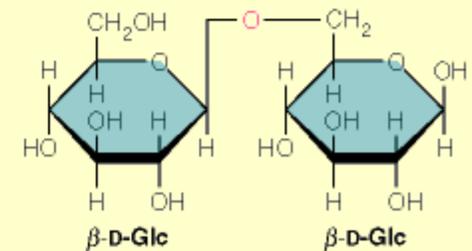
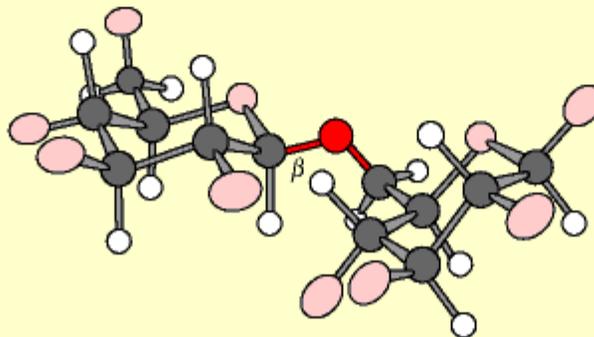
Cellobiose:
 β -D-glucopyranosyl
(1 \rightarrow 4) β -D-glucopyranose

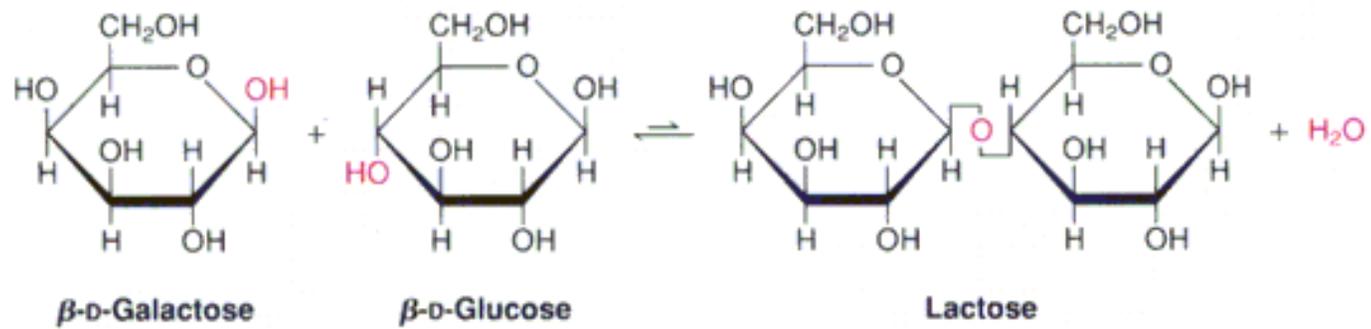


Lactose:
 β -D-galactopyranosyl
(1 \rightarrow 4) β -D-glucopyranose



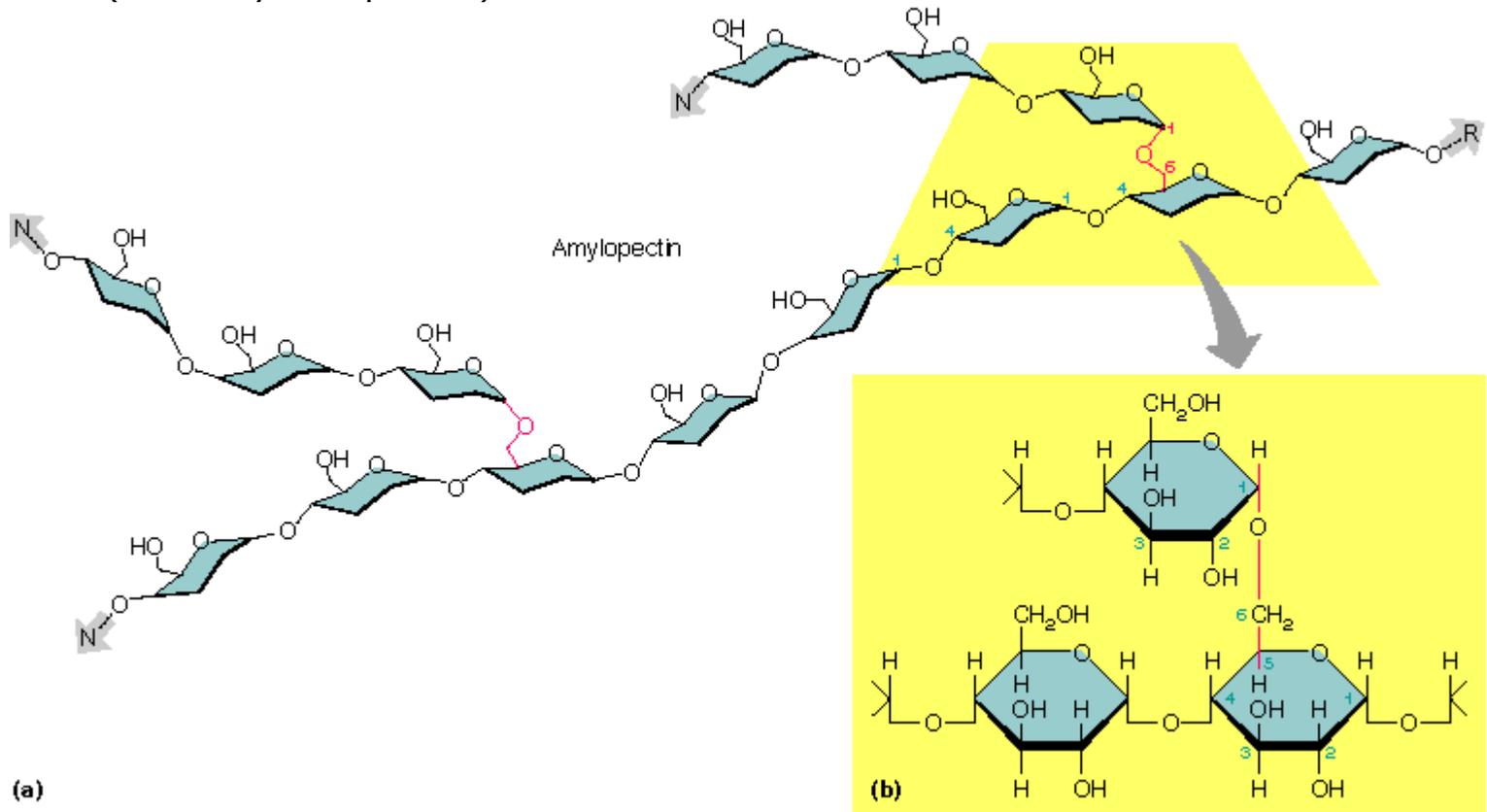
Gentiobiose:
 β -D-glucopyranosyl
(1 \rightarrow 6) β -D-glucopyranose



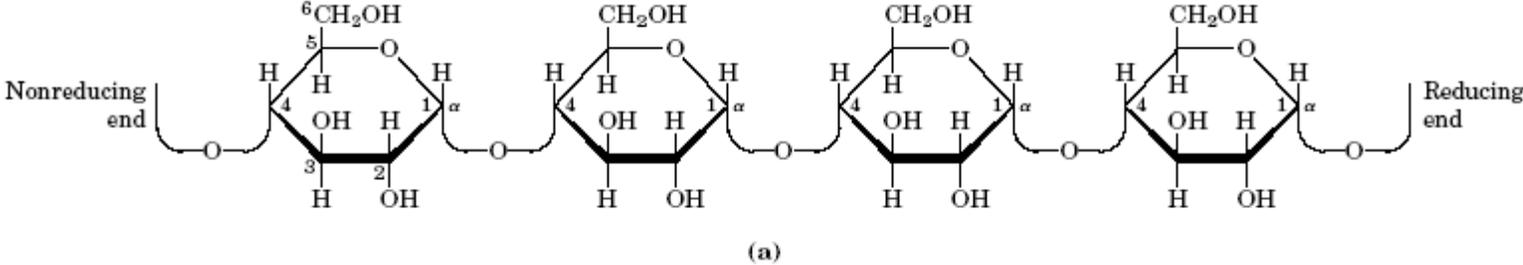


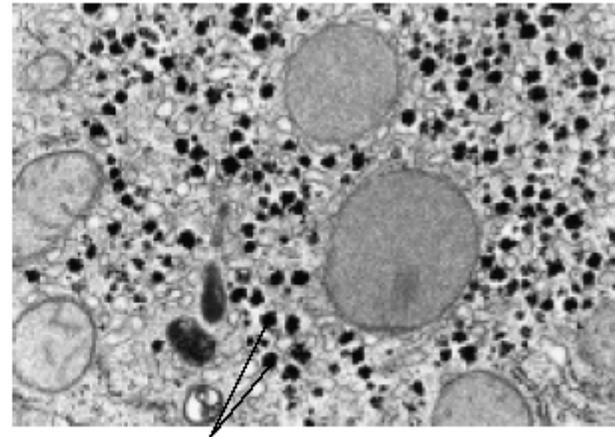
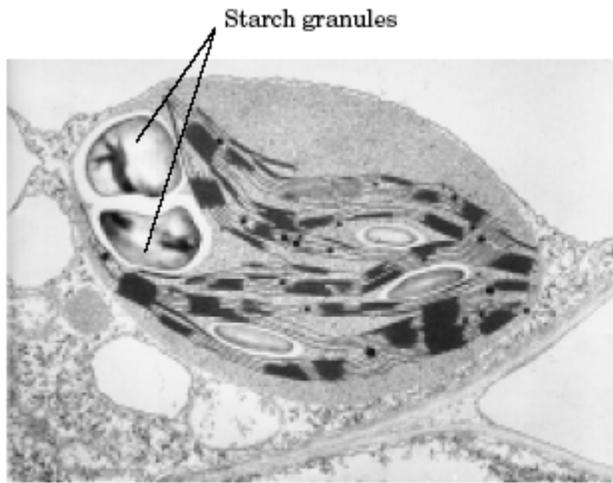
Glycanos (polisacáridos)

Almidón (amilosa y amilopectina)

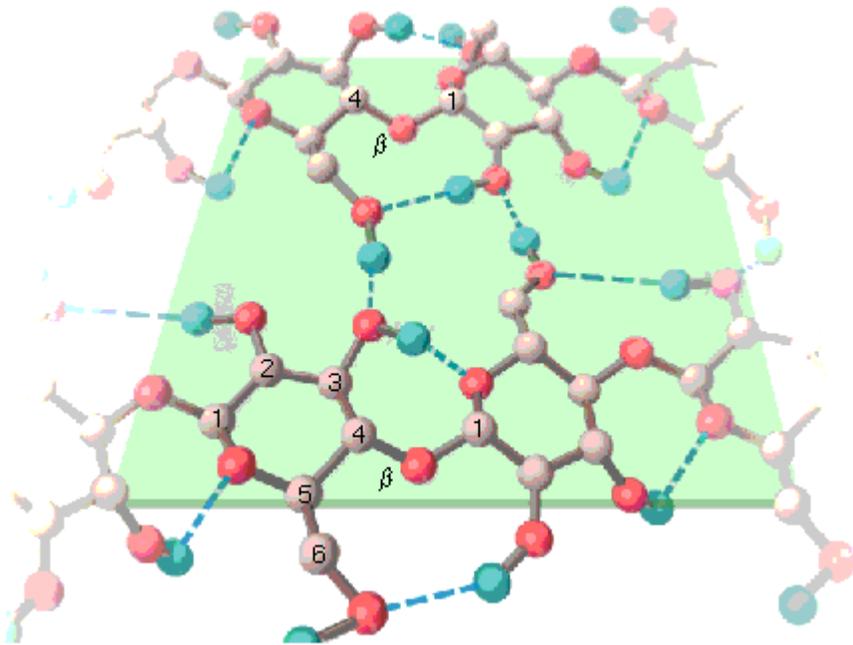


amilosa



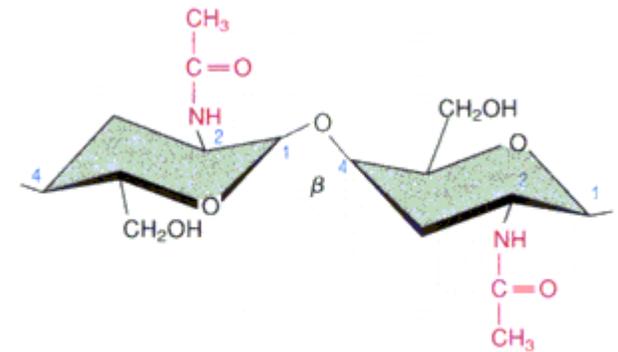


Alfa(1-4) y alfa(1-6)
poli D glucosa



Celulosa

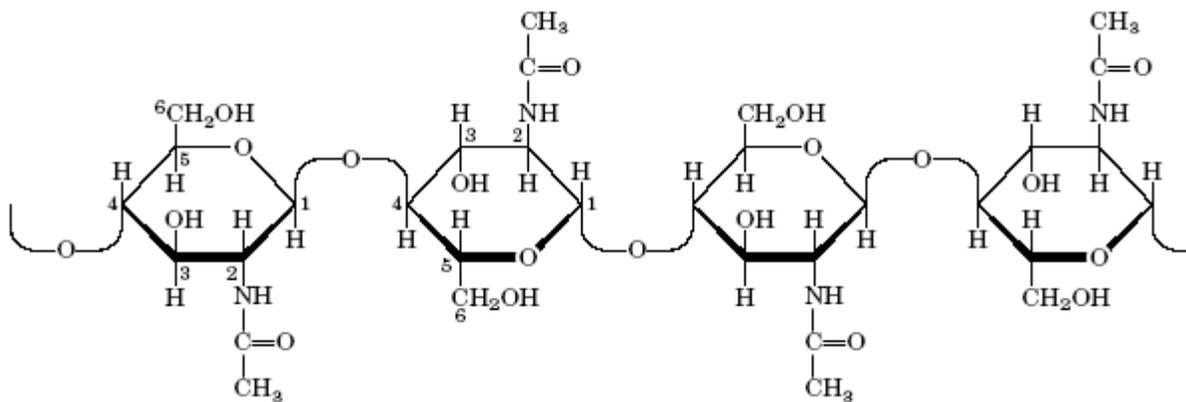
beta(1-4) pli D glucosa



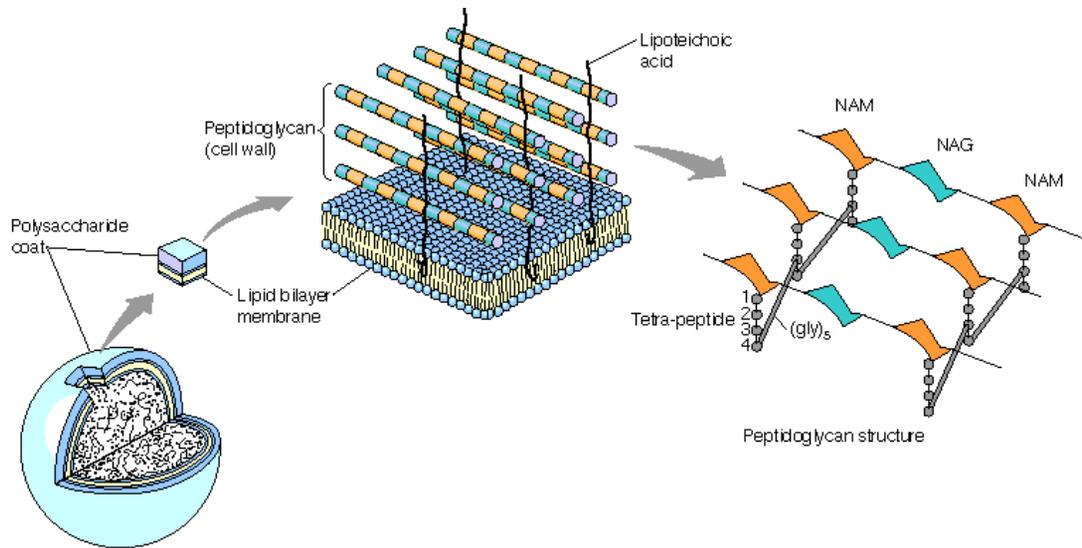
Chitin

Quitina

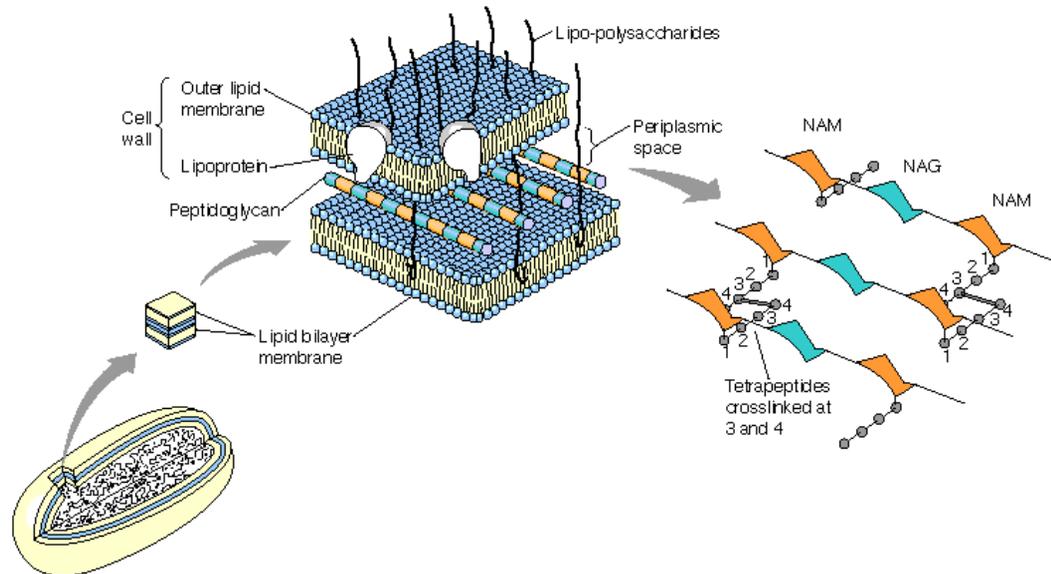
beta(1-4) poli D
Nacetilglucosamina



Peptidoglicano



(a) Gram positive:
Staphylococcus aureus



(b) Gram negative:
Escherichia coli

Glicosaminoglicanos

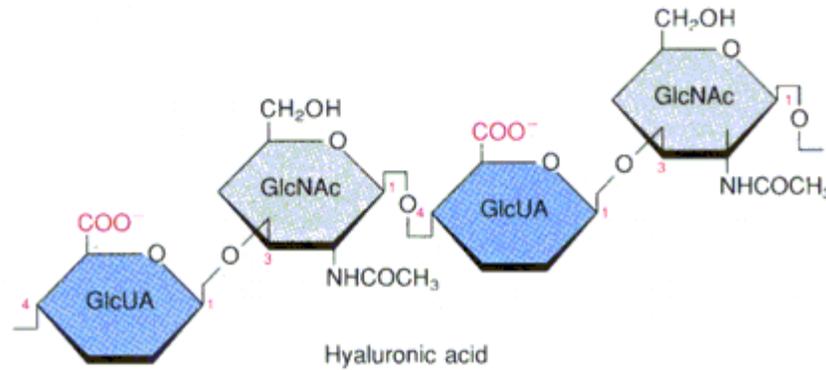
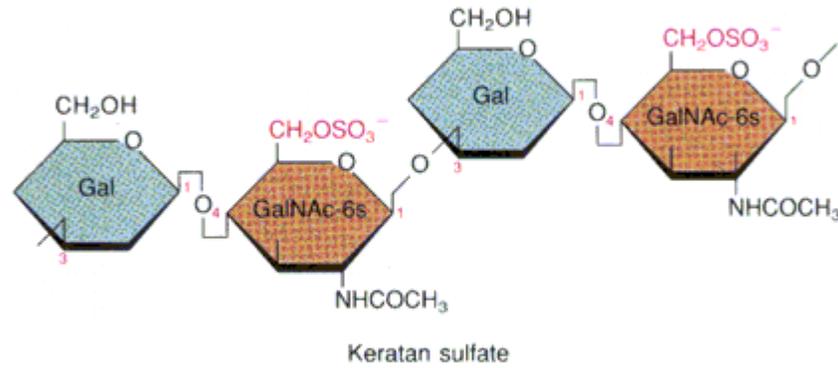


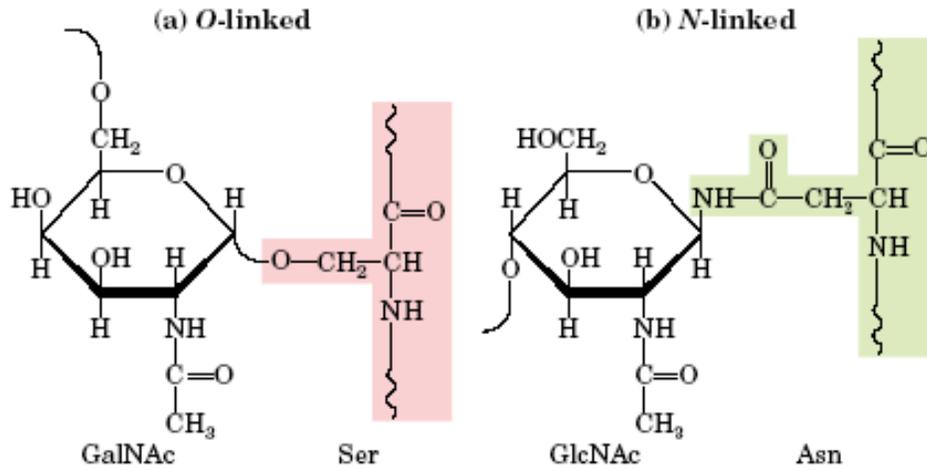
table 9-2

Structures and Roles of Some Polysaccharides				
Polymer	Type*	Repeating unit [†]	Size (number of monosaccharide units)	Roles
Starch				Energy storage: in plants
Amylose	Homo-	(α 1 \rightarrow 4)Glc, linear	50–5,000	
Amylopectin	Homo-	(α 1 \rightarrow 4)Glc, with (α 1 \rightarrow 6)Glc branches every 24 to 30 residues	Up to 10^6	
Glycogen	Homo-	(α 1 \rightarrow 4)Glc, with (α 1 \rightarrow 6)Glc branches every 8 to 12 residues	Up to 50,000	Energy storage: in bacteria and animal cells
Cellulose	Homo-	(β 1 \rightarrow 4)Glc	Up to 15,000	Structural: in plants, gives rigidity and strength to cell walls
Chitin	Homo-	(β 1 \rightarrow 4)GlcNAc	Very large	Structural: in insects, spiders, crustaceans, gives rigidity and strength to exoskeletons
Peptidoglycan	Hetero-; peptides attached	4)Mur2Ac(β 1 \rightarrow 4)GlcNAc(β 1	Very large	Structural: in bacteria, gives rigidity and strength to cell envelope
Hyaluronate (a glycosaminoglycan)	Hetero-; acidic	4)GlcA(β 1 \rightarrow 3)GlcNAc(β 1	Up to 100,000	Structural: in vertebrates, extracellular matrix of skin and connective tissue; viscosity and lubrication in joints

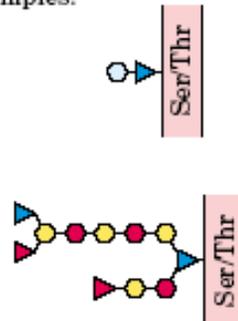
* Each polymer is classified as a homopolysaccharide (homo-) or heteropolysaccharide (hetero-).

[†]The abbreviated names for the peptidoglycan and hyaluronate repeating units indicate that the polymer contains repeats of this disaccharide unit, with the GlcNAc of one disaccharide unit linked β (1 \rightarrow 4) to the first residue of the next disaccharide unit.

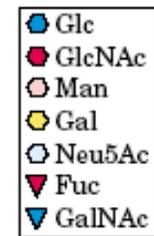
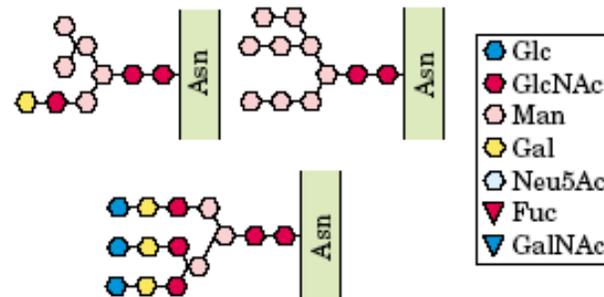
O y N-glicosidos. Glicoproteínas



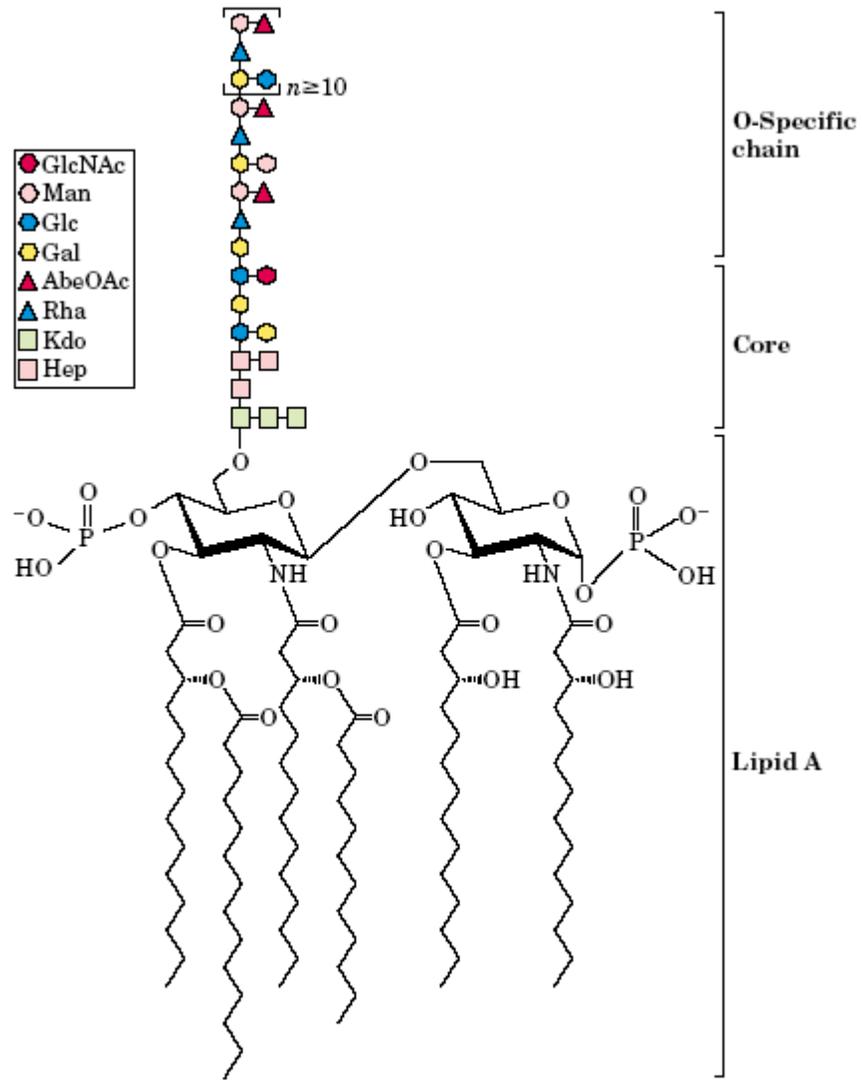
Examples:



Examples:



Lipopolisacáridos



(a)

